are described in a confusing manner.

of those engineers and managers in a wide range of activities involving LPG or LNG technology.

> ELISABETH M. DRAKE ARTHUR D. LITTLE, INC. Cambridge, Massachusetts 02140

Introduction to Chemical Engineering Analysis, T. W. Fraser Russell and Morton M. Denn, Wiley, New York (1972). 502 pages. \$17.75.

In the past two to three decades we have seen chemical engineering education swing from the Unit Operations-Technology school to the transport phenomena approach. At the extreme of this swing students have found the mathematics to be somewhat abstract and difficult to relate to reality. However, without this swing it is not easy to see how engineering could meet the demands of modern design. These authors present a good attempt to inject reality into the concepts of engineering analysis, which is the essence of engineering education.

Although the material covered is somewhat ambitious for an introductory text in chemical engineering, engiplace to begin the formal education of tions, along with the essentials of di-researchers. mensions and units. Since students oftroductory part of the text.

requires some maturity in mathematics, but Chapters 15, 16, and 17 of Part IV summarize the needed mathematics the impression that the social and ecovery effectively. The material in these nomic aspects are less well understood chapters will require some attention as it is needed in the other parts of the izing unwanted fires. text if the university does not have a strong applied mathematics depart- character of accidental fires, bringing ment. However, the use of mathe- in principal factors such as fuel loadmatics at this level should have a strong ing, geometry, and ventilation. The United States.) motivating influence on the students to chapter on fires in enclosures offers a

and energy balances in a refreshing with heat and mass transfer. Particuway. The order of the material may lar attention is paid to the condensed austenitic stainless steels, which is too require some instructors to reorient phase since the response of solid or high by about 150°C), typographical themselves, but the discussion of physi- liquid fuels in fires poses special analyti- errors (for example, on page 13, refcal and mathematical principles as they cal difficulties. Discussions of flames erence 41 should be 40 and on page 33, are needed in the text has proven to be and condensed phase interactions, mass in Table 3.2, the column headings are

are welcome additions to the libraries excellent introduction to graphical flames are also included. techniques. The instructor using this text will find a wide choice of examples lates textbook radiative heat transfer of engineering analysis, all very well theories to fires. The final two chapters chosen to illustrate the unique character of chemical engineering among the engineering disciplines.

ized an approach to introductory engineering education which this reviewer has found to be most effective through man injury and the generation of toxic fifteen years of cut-and-try. Modern engineering requires more engineering analysis than unit operations calculations of the old school, and students require an early motivation by seeing the utility of the material they study. This approach is a step in the direction which will satisfy both these needs.

> JAMES W. HALL Brown and Root, Inc. HOUSTON, TEXAS

Heat Transfer in Fires: Thermophysics, Social Aspects, Economic Impact, P. L. Blackshear (ed.), Scripta; Washington, D.C. (1974). 513 pages. \$28.50.

This is a collection of technical discourses by several authors which proneering analysis, as present in the first vides an excellent introduction to the three chapters of Part I, is an excellent state of the art of the combustion aspects of fire. With minor exceptions, engineers. The model development the various components of fire are dediagrams of Figures 3.5, 3.6, and 3.8 scribed in a well coordinated and copresent a concise picture of the thought hesive fashion. The fundamental relaprocesses used in the analysis of physitionships which have been developed cal processes. This section places to date are reviewed and with the refproper emphasis on the mathematical erences cited, this book provides a with the technical terms and usage model as a foundation for later sec-comprehensive source for engineers and normal to our educational background

ten question the need for courses in cial and economic implications of fire. chemistry, good choice is made of ex- It presents reasons to better understand amples in reaction kinetics in this in- the physics of fire. The annual losses Rt, the UTS at design temperature. The approach used in this textbook question of how much should be spent its own terminology; for example, units expect, however, the reader is left with than the growing science of character-

The main text discusses the gross master the topics covered in typical concise summary of spreading, flash-sufficient to recommend against this mathematics courses.

over, and fully-developed fires. This is book, it also suffers from inaccuracies Parts II and III introduce the mass followed by several chapters dealing

gineering education. The use of data of pyrolysis encompass the principal On balance, nevertheless, both books analysis in the examples serves as an phenomena. The fluid mechanics of

> The next section of the book reprovide masterful summaries of fire spread and ignition.

Although the reader can conceive of The authors of this text have formal- other phenomena which are important parts of the fire problem, (for example, heat transfer processes relating to hugases), and which might be expected to fall within the implied objectives of the book, this work does serve as a very valuable reference for technicians concerned with accidental fires.

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Selecting Engineering Materials for Chemical and Process Plant, L. S. Evans, Wiley, New York-Toronto (1974). 164 pages. \$14.95.

This book purports to be of value to "chemical, design and maintenance engineers responsible for the choice of engineering materials, in all process in-dustries" and to be "suited to all material science chemical engineering and mechanical engineering courses in universities". This reviewer feels that only those individuals who have recently emigrated from the United Kingdom could find it comfortable to use. For use in America, it can only be confusing and strange, certainly inconsistent and technological literature. For ex-The first section focuses on the so- ample, "E" is the symbol for yield strength, R₂₀, the room temperature ultimate tensile strength (UTS) and from fire are cited to introduce the Further, the book is inconsistent in to reduce these losses. As one might for stress range from N/mm.2 and N/cm.² to KN/m², while pressure is given both in Kgf/cm.² and N/cm². (A conversion table might prove of value to those not conversant with the mental gymnastics required to cope with so wide a range of units and to help anchor them to the American-English system still current in the

If the foregoing drawbacks were in-(for example, 600°C is cited as the temperature for the onset of creep in an effective approach in modern en- and energy balances, and the kinetics misprinted), and careless printing (for